

- BRADY, O. L. (1914). *J. Chem. Soc.* pp. 2104–2116.
- BUSING, W. R., MARTIN, K. O. & LEVY, H. A. (1962). ORFLS. Report ORNL-TM-305. Oak Ridge National Laboratory, Tennessee.
- CRAVEN, B. M. & McMULLAN, R. K. (1979). *Acta Cryst. B35*, 934–945.
- CYVIN, S. J. (1968). *Molecular Vibrations and Mean Square Amplitudes*. Amsterdam: Elsevier.
- DEFREES, D. J. (1980). Unpublished data.
- DEFREES, D. J., LEVI, B. A., POLLACK, S. K., HEHRE, W. J., BINKLEY, J. S. & POPLE, J. A. (1979). *J. Am. Chem. Soc.* **101**, 4085–4089.
- FLORA, K. P., VAN'T REIT, B. & WAMPLER, G. L. (1978). *Cancer Res.* **38**, 1291–1295.
- HALL, D. (1965). *Acta Cryst.* **18**, 955–958.
- HALL, D. & LLEWELLYN, F. J. (1956). *Acta Cryst.* **9**, 108–112.
- HARIHARAN, P. C. & POPLE, J. A. (1974). *Mol. Phys.* **27**, 209–214.
- HARMONY, M. D., LAURIE, V. W., KUCZKOWSKI, R. L., SCHWENDEMAN, R. H., RAMSAY, D. A., LOVAS, F. J., LAFFERTY, W. J. & MAKI, A. G. (1979). *J. Phys. Chem. Ref. Data*, **8**, 619–721.
- HAWKES, G. E., HERWIG, K. & ROBERTS, J. D. (1974). *J. Org. Chem.* **39**, 1017–1028.
- HEHRE, W. J., STEWART, R. F. & POPLE, J. A. (1969). *J. Chem. Phys.* **51**, 2657–2664.
- HUTCHINGS, M. T., SHULHOF, M. P. & GUGGENHEIM, H. J. (1972). *Phys. Rev. B*, **5**, 154–168.
- JEFFREY, G. A., RUBLE, J. R., McMULLAN, R. K., DEFREES, D. J., BINKLEY, J. S. & POPLE, J. A. (1980). *Acta Cryst. B36*, 2292–2299.
- JOHNSON, C. K. (1970). *Thermal Neutron Diffraction*, edited by B. T. M. WILLIS, ch. 9, pp. 132–159. Oxford Univ. Press.
- JOHNSON, C. K. (1976). ORTEP II. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee.
- KLEINSPEHN, G. G., JUNG, J. A. & STUDNIARZ, S. A. (1967). *J. Org. Chem.* **32**, 460–462.
- KOESTER, L. (1977). *Springer Tracts in Modern Physics*. Vol. 80, edited by G. HOHLER, p. 37. New York: Springer-Verlag.
- MCMULLAN, R. K. & KOETZLE, T. (1980). Unpublished.
- MEULENAER, J. DE & TOMPA, H. (1965). *Acta Cryst.* **19**, 1014–1018.
- MØLLER, C. & PLESSET, M. S. (1934). *Phys. Rev.* **46**, 618–622.
- OLVSSON, I. & JÖNSSON, P.-G. (1976). *The Hydrogen Bond – Recent Developments in Theory and Experiments*, edited by P. SCHUSTER. Amsterdam: North-Holland.
- ORVILLE-THOMAS, W. J. & PARSONS, A. E. (1958). *Trans. Faraday Soc.* **54**, 460–468.
- POPLE, J. A., BINKLEY, J. S. & SEEGER, R. (1976). *Int. J. Quantum Chem. Symp.* **10**, 1–19.
- POPLE, J. A., KRISHNAN, R., SCHLEGEL, H. B. & BINKLEY, J. S. (1979). *Int. J. Quantum Chem. Symp.* **13**, 225–241.
- ROSENKRANZ, H. S., HJORTH, R. & CARR, H. S. (1971). *Biochim. Biophys. Acta*, **232**, 48–60.
- TEMPLETON, L. K. & TEMPLETON, D. H. (1973). Am. Crystallogr. Assoc. Meeting, Storrs, CT. Abstract E10.
- WHITESIDE, R. A., BINKLEY, J. S., KRISHNAN, R., DEFREES, D. J., SCHLEGEL, H. B. & POPLE, J. A. (1980). Carnegie-Mellon Quantum Chemistry Archive. Chemistry Department, Carnegie-Mellon Univ., Pittsburgh, PA 15213.

Acta Cryst. (1981). **B37**, 1387–1390

cis-cisoid-cis-transoid-cis-cisoid-cis-Perhydro-cis-6,13-pentacenediyi Diacetate

BY P. VAN NUFFEL, G. H. PETIT, A. T. H. LENSTRA AND J. VAN LOOCK

University of Antwerp (UIA), Department of Chemistry, Universiteitsplein 1, B-2610 Wilrijk, Belgium

AND P. VANHEE

Department of Organic Chemistry, State University Gent, Krijgslaan 271 (S-4bis), B-9000 Gent, Belgium

(Received 16 December 1980; accepted 21 January 1981)

Abstract

$C_{26}H_{40}O_4$ is orthorhombic, *Pbca*, with $Z = 8$, $a = 9.972(4)$, $b = 21.285(4)$, $c = 21.825(9)$ Å, $D_c = 1.195$ Mg m $^{-3}$. $R_w = 0.035$ for 2160 reflections. Ring *B* occurs in a boat-like conformation, while the other four six-membered rings are in a distorted chair conformation.

0567-7408/81/071387-04\$01.00

Introduction

The title compound, shown in Fig. 1, is important in a study of the ring inversion in perhydropolycenes (De Pessenier, Anteunis & Tavernier, 1978; Vanhee, De Pessenier, Anteunis & Tavernier, 1979). It was first synthesized by Vanhee (1981). A determination of its configuration from spectroscopic data only was difficult

© 1981 International Union of Crystallography

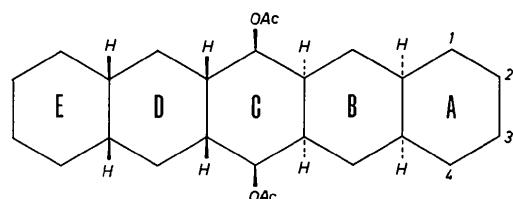


Fig. 1. Structural formula with nomenclature of the rings.

because of the complexity of the structure, the number of possible configurations (ten asymmetric C atoms) and the fact that the NMR investigation had shown the existence of two conformers, each present for about 50%. Therefore an X-ray investigation was undertaken to establish the configuration as well as the conformation in the solid state.

3207 independent reflections ($0 < \theta < 23^\circ$) were measured at room temperature on an Enraf–Nonius CAD-4 automated diffractometer with Zr-filtered Mo $K\alpha$ radiation. A pure ω scan was used. Intensities were corrected for Lorentz and polarization factors. In view of the low absorption coefficient ($\mu = 0.0847 \text{ mm}^{-1}$) and the size of the crystal ($0.2 \times 0.2 \times 0.25 \text{ mm}$) no absorption correction was made.

The structure was solved with *MULTAN* (Germain, Main & Woolfson, 1971). An *E* map using terms with $E > 1.6$ revealed all non-hydrogen atoms. After refinement a difference map showed the positions of most of the H atoms; the five remaining H atoms were placed at expected positions.

The isotropic temperature factors of the H atoms attached to the ring system were fixed at $B_{\text{iso}} = 4.0 \text{ \AA}^2$, those of the acetyl groups at 5.0 \AA^2 . The structure was refined by block-diagonal least squares to a final $R_w = 0.035$ {defined by $\{\sum w(|F_o| - |F_c|)^2 / \sum w|F_o|^2\}^{1/2}$ } for 2160 reflections with $I > 2.0\sigma(I)$.

A weighting scheme based on counting statistics was employed. The final difference map showed no maxima $> 0.1 \text{ e \AA}^{-3}$. The atomic parameters are presented in Table 1. Fig. 2 shows the conformation of the molecule.*

Discussion

The first result of this work is the assessment of the *cis-cisoid-cis-transoid-cis-cisoid-cis* positioning of the rings *A–E* and of the *cis* orientation of the acetyl groups, confirming the tentative conclusions about the configuration drawn from NMR data.

Bond distances and angles are presented in Tables 2 and 3. The large spread in C–C distances (1.510–

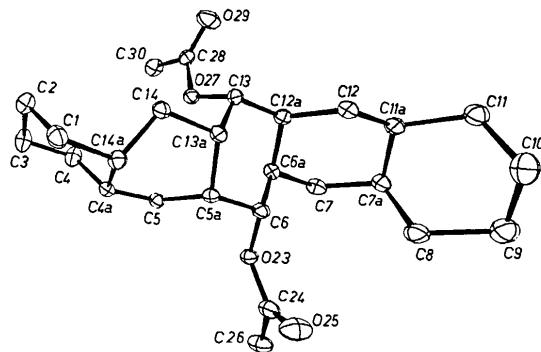


Fig. 2. The conformation of the title compound and the numbering of the atoms.

1.551 Å) and CCC angles (109.5–116.9°) is not unusual. It reflects the non-equivalence of the various sp^3 – sp^3 bonds. Similar phenomena were established in the perhydrophenanthrene skeleton (Romers, Altona, Jacobs & De Graaff, 1974). No significant differences are observed between the equatorially oriented acetyl group on C(6) and the axial one on C(13). The orientation of the acetyl groups differs considerably from the usual *gauche* position: torsion angles C(5a)–C(6)–O(23)–C(24) and C(12a)–C(13)–O(27)–C(28) are –93.2 and –89.8°, respectively.

The title compound has an unusual conformation. Rings *A*, *C*, *D* and *E* occur in a distorted chair form, but ring *B* takes on a boat-like conformation. At first sight this may seem surprising. A Dreiding model of the all-chair conformation, however, shows severe steric hindrance between ring *A* and the axial acetyl group on C(13). This steric compression cannot be released by converting ring *C* into a twist-boat form, but it can be diminished by converting ring *B* into a boat-like conformation. Furthermore, the energy difference between the non-substituted all-chair conformation and the form in which ring *B* has a boat-like conformation is not expected to be large. The arguments for this proposition stem from results of molecular mechanics calculations on *cis-cisoid-cis*-perhydroanthracene. These results may to some extent be extrapolated to perhydropentacene, since the latter can be considered as a combination of two perhydroanthracene units. The calculations (Allinger & Wuesthoff, 1971) show that in *cis-cisoid-cis*-perhydroanthracene the chair–chair–chair form is only 8.8 kJ mol^{–1} lower in energy than the chair–boat–chair form. The occurrence of a flexible form of ring *B* is a strong indication that the temperature dependence of the NMR spectra of the title compound is correctly attributed to a conformational equilibrium within ring *B* in the liquid state.

The ring torsion angles, Table 4, indicate that the rings are flattened with respect to an ideal six-membered ring. This flattening is undoubtedly introduced to decrease the unfavourable axial–axial 1,3 interactions, such as C(6)…C(8) in ring *D* and

* Lists of structure factors and anisotropic thermal parameters have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 35935 (14 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

Table 1. Positional parameters in fractions of cell edges

The e.s.d.'s given in parentheses refer to the last significant digit. Isotropic temperature factors (\AA^2) of non-hydrogen atoms are calculated from anisotropic temperature parameters assuming equal volume of the 50% probability region. All anisotropic thermal parameters were physically acceptable. $H(xj)$ ($j = 1, 2, 3$) is attached to $C(x)$, $H(xa)$ to atom $C(xa)$. B_{iso} was calculated according to Lipson & Cochran (1968): $B_{\text{iso}} = 8\pi^2(U_{11}^a U_{22}^a U_{33}^a)^{1/3}$.

	<i>x</i>	<i>y</i>	<i>z</i>	B_{iso} (\AA^2)
C(1)	-0.2225 (3)	0.0014 (2)	0.3621 (1)	4.32 (14)
C(2)	-0.1927 (3)	0.0708 (2)	0.3531 (1)	4.68 (14)
C(3)	-0.1256 (3)	0.0809 (1)	0.2919 (1)	4.91 (13)
C(4)	0.0016 (3)	0.0426 (1)	0.2879 (1)	4.46 (13)
C(4a)	-0.0207 (3)	-0.0279 (1)	0.2976 (1)	3.43 (15)
C(5)	0.1143 (3)	-0.0640 (1)	0.2953 (1)	3.19 (14)
C(5a)	0.1260 (3)	-0.1126 (1)	0.3461 (1)	2.81 (14)
C(6)	0.2507 (3)	-0.1549 (1)	0.3450 (1)	3.06 (13)
C(6a)	0.3797 (3)	-0.1257 (1)	0.3692 (1)	2.96 (13)
C(7)	0.5019 (3)	-0.1698 (1)	0.3692 (1)	3.88 (15)
C(7a)	0.5063 (3)	-0.2209 (1)	0.4194 (1)	3.46 (14)
C(8)	0.4247 (3)	-0.2797 (1)	0.4056 (1)	4.37 (18)
C(9)	0.4340 (3)	-0.3282 (1)	0.4563 (1)	4.46 (15)
C(10)	0.3870 (3)	-0.2997 (1)	0.5162 (1)	4.34 (13)
C(11)	0.4708 (3)	-0.2426 (1)	0.5329 (1)	3.94 (12)
C(11a)	0.4743 (3)	-0.1927 (1)	0.4827 (1)	3.37 (16)
C(12)	0.3472 (3)	-0.1526 (1)	0.4814 (1)	3.19 (15)
C(12a)	0.3575 (3)	-0.0998 (1)	0.4342 (1)	3.05 (17)
C(13)	0.2382 (3)	-0.0553 (1)	0.4360 (1)	3.07 (12)
C(13a)	0.1101 (3)	-0.0831 (1)	0.4101 (1)	2.87 (13)
C(14)	-0.0073 (3)	-0.0359 (1)	0.4123 (1)	3.40 (14)
C(14a)	-0.0993 (3)	-0.0405 (1)	0.3572 (1)	3.37 (12)
O(23)	0.2824 (2)	-0.17499 (8)	0.28235 (7)	3.48 (11)
C(24)	0.2321 (3)	-0.2302 (1)	0.2629 (1)	3.98 (18)
O(25)	0.1511 (2)	-0.2604 (1)	0.2906 (1)	5.83 (19)
C(26)	0.2940 (3)	-0.2472 (2)	0.2026 (1)	5.09 (25)
O(27)	0.2747 (2)	0.00013 (9)	0.39923 (8)	3.53 (17)
C(28)	0.3352 (3)	0.0480 (1)	0.4278 (1)	3.88 (17)
O(29)	0.3496 (3)	0.0505 (1)	0.48218 (9)	6.00 (22)
C(30)	0.3773 (3)	0.0969 (2)	0.3836 (2)	5.40 (24)
H(11)	-0.265 (2)	-0.005 (1)	0.4015 (9)	4.00
H(12)	-0.299 (2)	-0.011 (1)	0.329 (1)	4.00
H(21)	-0.280 (2)	0.093 (1)	0.3540 (9)	4.00
H(22)	-0.131 (2)	0.087 (1)	0.3910 (9)	4.00
H(31)	-0.110 (2)	0.129 (1)	0.2845 (9)	4.00
H(32)	-0.178 (2)	0.071 (1)	0.259 (1)	4.00
H(41)	0.077 (2)	0.056 (1)	0.322 (1)	4.00
H(42)	0.056 (2)	0.047 (1)	0.253 (1)	4.00
H(4a)	-0.083 (2)	-0.042 (1)	0.263 (1)	4.00
H(51)	0.127 (2)	-0.0853 (9)	0.255 (1)	4.00
H(52)	0.176 (2)	-0.036 (1)	0.2982 (9)	4.00
H(5a)	0.059 (2)	-0.141 (1)	0.3418 (9)	4.00
H(61)	0.224 (2)	-0.191 (1)	0.3682 (9)	4.00
H(6a)	0.397 (2)	-0.089 (1)	0.341 (1)	4.00
H(71)	0.512 (2)	-0.191 (1)	0.330 (1)	4.00
H(72)	0.593 (2)	-0.144 (1)	0.3743 (9)	4.00
H(7a)	0.604 (2)	-0.237 (1)	0.422 (1)	4.00
H(81)	0.454 (2)	-0.296 (1)	0.3669 (9)	4.00
H(82)	0.344 (2)	-0.272 (1)	0.3994 (9)	4.00
H(91)	0.539 (2)	-0.341 (1)	0.4610 (9)	4.00
H(92)	0.376 (2)	-0.370 (1)	0.449 (1)	4.00
H(101)	0.309 (2)	-0.289 (1)	0.510 (1)	4.00
H(102)	0.385 (2)	-0.332 (1)	0.5509 (9)	4.00
H(111)	0.568 (2)	-0.263 (1)	0.5378 (9)	4.00
H(112)	0.428 (2)	-0.225 (1)	0.570 (1)	4.00
H(11a)	0.558 (2)	-0.163 (1)	0.4922 (9)	4.00
H(121)	0.330 (2)	-0.134 (1)	0.5225 (9)	4.00

Table 1 (cont.)

	<i>x</i>	<i>y</i>	<i>z</i>	B_{iso} (\AA^2)
H(122)	0.266 (2)	-0.177 (1)	0.4747 (9)	4.00
H(12a)	0.428 (2)	-0.075 (1)	0.442 (1)	4.00
H(131)	0.212 (2)	-0.041 (1)	0.4792 (9)	4.00
H(13a)	0.095 (2)	-0.118 (1)	0.441 (1)	4.00
H(141)	0.027 (2)	0.008 (1)	0.417 (1)	4.00
H(142)	-0.055 (2)	-0.040 (1)	0.4506 (9)	4.00
H(14a)	-0.133 (2)	-0.084 (1)	0.3556 (9)	4.00
H(261)	0.240 (2)	-0.278 (1)	0.183 (1)	5.00
H(262)	0.377 (2)	-0.271 (1)	0.212 (1)	5.00
H(263)	0.318 (2)	-0.211 (1)	0.177 (1)	5.00
H(301)	0.385 (2)	-0.138 (1)	0.401 (1)	5.00
H(302)	0.442 (2)	0.088 (1)	0.371 (1)	5.00
H(303)	0.309 (2)	0.107 (1)	0.352 (1)	5.00

Table 2. Bond distances (\AA) with e.s.d.'s in parentheses

C(1)–C(2)	1.520 (6)	C(8)–C(9)	1.515 (6)
C(2)–C(3)	1.510 (6)	C(9)–C(10)	1.515 (6)
C(3)–C(4)	1.511 (6)	C(10)–C(11)	1.520 (6)
C(4)–C(4a)	1.530 (6)	C(11)–C(11a)	1.525 (4)
C(4a)–C(5)	1.551 (6)	C(11a)–C(12)	1.528 (4)
C(5)–C(5a)	1.521 (6)	C(12)–C(12a)	1.528 (4)
C(5a)–C(6)	1.534 (4)	C(12a)–C(13)	1.520 (4)
C(6)–C(6a)	1.523 (4)	C(13)–C(13a)	1.517 (4)
C(6a)–C(7)	1.538 (6)	C(13a)–C(14)	1.544 (6)
C(7)–C(7a)	1.544 (6)	C(14)–C(14a)	1.517 (4)
C(7a)–C(8)	1.524 (6)	C(14a)–C(1)	1.522 (6)
C(4a)–C(14a)	1.542 (4)	C(6a)–C(12a)	1.539 (4)
C(5a)–C(13a)	1.540 (4)	C(7a)–C(11a)	1.541 (4)
C(6)–O(23)	1.467 (4)	C(13)–O(27)	1.473 (4)
O(23)–C(24)	1.347 (4)	O(27)–C(28)	1.338 (4)
C(24)=O(25)	1.196 (4)	C(28)=O(29)	1.196 (4)
C(24)–C(26)	1.499 (6)	C(28)–C(30)	1.481 (6)
Average C–H bond		0.98 (3)	

Table 3. Bond angles ($^\circ$) with e.s.d.'s in parentheses

C(2)–C(1)–C(14a)	113.7 (4)	C(7a)–C(8)–C(9)	112.5 (4)
C(1)–C(2)–C(3)	109.8 (4)	C(8)–C(9)–C(10)	109.8 (4)
C(2)–C(3)–C(4)	110.3 (4)	C(9)–C(10)–C(11)	111.0 (4)
C(3)–C(4)–C(4a)	113.6 (4)	C(10)–C(11)–C(11a)	113.4 (4)
C(4)–C(4a)–C(5)	110.8 (3)	C(7a)–C(11a)–C(11)	112.2 (3)
C(4)–C(4a)–C(14a)	111.1 (3)	C(7a)–C(11a)–C(12)	111.9 (3)
C(5)–C(4a)–C(14a)	112.5 (3)	C(11)–C(11a)–C(12)	112.6 (3)
C(4a)–C(5)–C(5a)	112.4 (3)	C(11a)–C(12)–C(12a)	111.6 (3)
C(5)–C(5a)–C(6)	116.7 (3)	C(6a)–C(12a)–C(12)	111.6 (3)
C(5)–C(5a)–C(13a)	112.1 (3)	C(6a)–C(12a)–C(13)	111.1 (3)
C(6)–C(5a)–C(13a)	109.7 (3)	C(12)–C(12a)–C(13)	112.8 (3)
C(5a)–C(6)–C(6a)	116.1 (3)	C(12a)–C(13)–C(13a)	114.0 (3)
C(6)–C(6a)–C(7)	114.9 (3)	C(5a)–C(13a)–C(13)	114.3 (3)
C(6)–C(6a)–C(12a)	110.2 (3)	C(5a)–C(13a)–C(14)	111.9 (3)
C(7)–C(6a)–C(12a)	109.5 (3)	C(13)–C(13a)–C(14)	111.9 (3)
C(6a)–C(7)–C(7a)	116.9 (3)	C(13a)–C(14)–C(14a)	113.0 (3)
C(7)–C(7a)–C(8)	115.1 (4)	C(1)–C(14a)–C(4a)	111.6 (3)
C(7)–C(7a)–C(11a)	110.9 (3)	C(1)–C(14a)–C(14)	113.2 (3)
C(8)–C(7a)–C(11a)	112.7 (4)	C(4a)–C(14a)–C(14)	110.5 (3)
C(5a)–C(6)–O(23)	111.1 (3)	C(12a)–C(13)–O(27)	107.0 (3)
C(6a)–C(6)–O(23)	105.1 (3)	C(13a)–C(13)–O(27)	108.5 (3)
C(6)–O(23)–C(24)	117.9 (3)	C(13)–O(27)–C(28)	117.8 (3)
O(23)–C(24)=O(25)	124.1 (4)	O(27)–C(28)=O(29)	123.5 (4)
O(23)–C(24)–C(26)	109.5 (4)	O(27)–C(28)–C(30)	111.0 (4)
O(25)=C(24)–C(26)	126.4 (4)	O(27)–C(28)–C(30)	125.5 (4)
Average H–C–H angle		106 (3)	
Average C–C–H angle		109 (2)	

Table 4. Ring torsion angles (°) with e.s.d.'s in parentheses

Ring A		Ring C		Ring E	
C(14a)—C(1)—C(2)—C(3)	56.6 (7)	C(13a)—C(5a)—C(6)—C(6a)	-50.1 (7)	C(11a)—C(7a)—C(8)—C(9)	-51.9 (7)
C(1)—C(2)—C(3)—C(4)	-57.7 (7)	C(5a)—C(6)—C(6a)—C(12a)	54.2 (7)	C(7a)—C(8)—C(9)—C(10)	57.8 (7)
C(2)—C(3)—C(4)—C(4a)	57.4 (7)	C(6)—C(6a)—C(12a)—C(13)	-53.4 (7)	C(8)—C(9)—C(10)—C(11)	-58.7 (7)
C(3)—C(4)—C(4a)—C(14a)	-52.1 (7)	C(6a)—C(12a)—C(13)—C(13a)	53.3 (7)	C(9)—C(10)—C(11)—C(11a)	55.2 (7)
C(4)—C(4a)—C(14a)—C(1)	48.0 (7)	C(12a)—C(13)—C(13a)—C(5a)	-50.9 (7)	C(10)—C(11)—C(11a)—C(7a)	-48.5 (7)
C(4a)—C(14a)—C(1)—C(2)	-51.8 (7)	C(13)—C(13a)—C(5a)—C(6)	46.9 (7)	C(11)—C(11a)—C(7a)—C(8)	46.4 (7)
Ring B		Ring D			
C(14a)—C(4a)—C(5)—C(5a)	10.8 (7)	C(12a)—C(6a)—C(7)—C(7a)	49.4 (7)		
C(4a)—C(5)—C(5a)—C(13a)	-57.2 (7)	C(6a)—C(7)—C(7a)—C(11a)	-47.5 (7)		
C(5)—C(5a)—C(13a)—C(14)	44.0 (7)	C(7)—C(7a)—C(11a)—C(12)	49.5 (7)		
C(5a)—C(13a)—C(14)—C(14a)	13.7 (7)	C(7a)—C(11a)—C(12)—C(12a)	-57.1 (7)		
C(13a)—C(14)—C(14a)—C(4a)	-59.7 (7)	C(11a)—C(12)—C(12a)—C(6a)	59.6 (7)		
C(14)—C(14a)—C(4a)—C(5)	46.1 (7)	C(12)—C(12a)—C(6a)—C(7)	-53.8 (7)		

Table 5. Orientation of the substituents on the various rings, in terms of the polar angle α (e.s.d.'s 0.7°)

Ring system	Substituent	On atom	Orientation	α (°)
Ring A	C(5)	C(4a)	Equatorial	21.4
	C(14)	C(14a)	Axial	76.4
Ring B	C(4)	C(4a)	Axial	60.6
	C(6)	C(5a)	Equatorial	-1.0
	C(13)	C(13a)	Inclinal	56.2
	C(1)	C(14a)	Equatorial	1.3
Ring C	C(5)	C(5a)	Axial	70.7
	O(23)	C(6)	Equatorial	22.3
	C(7)	C(6a)	Equatorial	-17.3
	C(12)	C(12a)	Axial	-80.3
	O(27)	C(13)	Axial	86.1
	C(14)	C(13a)	Equatorial	18.6
Ring D	C(6)	C(6a)	Axial	-76.7
	C(8)	C(7a)	Axial	-70.6
	C(11)	C(11a)	Equatorial	-15.6
	C(13)	C(12a)	Equatorial	-13.5
Ring E	C(7)	C(7a)	Equatorial	-17.5
	C(12)	C(11a)	Axial	-73.2

Table 6. Ring-puckering coordinates of the rings A-E

For a definition and the way they are calculated see Cremer & Pople (1975) and Cremer (1975).

Ring	q_2 (Å)	q_3 (Å)	Q (Å)	θ (°)	ϕ (°)
A	0.056	-0.547	0.550	174.2	273.3
B	0.733	0.012	0.733	89.10	288.5
C	0.035	0.515	0.516	3.94	138.3
D	0.076	-0.539	0.544	172.0	81.4
E	0.078	0.538	0.544	8.3	133.8
Ideal chair	0	0.560	0.560	0 180}	-
Ideal boat	0.7	0	0.7	90	$n\pi/3$

C(5)···O(27) in ring C. These and similar interactions can best be seen from Table 5, which gives a summary of the positions of the substituents on the individual rings in terms of the polar angle α ; α is the angle between a substituent and the mean plane of the ring

(Cremer & Pople, 1975). Substituents for which $\alpha > 0$ are above, those for which $\alpha < 0$ are below the plane of the ring. An orientation is considered equatorial when $0 \leq |\alpha| \leq 30^\circ$, inclinal when $30 < |\alpha| < 60^\circ$ and axial when $60 \leq |\alpha| \leq 90^\circ$. Other meaningful observables of ring geometries are the ring-puckering coordinates listed in Table 6. In comparison with an unsubstituted cyclohexane ring, the values found for the title compound show that rings A-E are rather distorted.

Finally, the value $\varphi(\text{ring } B) = 288.5^\circ$ indicates that ring B is in a twist form nearest to a true boat with C(5a) and C(14a) as bowsprits.

Two of us (PVN and GHP) thank the Belgian Organization for the Advancement of Research in Industry and Agriculture (IWONL) for a predoctoral grant. The authors are indebted to Professor Dr M. Anteunis (Gent), who suggested the subject and to Professor Dr H. Geise (Antwerp) for helpful discussions.

References

- ALLINGER, N. L. & WUESTHOFF, M. T. (1971). *J. Org. Chem.* **36**, 2051–2053.
- CREMER, D. (1975). *Quantum Chemistry Program Exchange*, **11**, 288.
- CREMER, D. & POPEL, J. A. (1975). *J. Am. Chem. Soc.* **97**, 1354–1358.
- DE PESSENIER, F., ANTEUNIS, M. & TAVERNIER, D. (1978). *Bull. Soc. Chim. Belg.* **87**, 179–187.
- GERMAIN, G., MAIN, P. & WOOLFSON, M. M. (1971). *Acta Cryst. A* **27**, 368–376.
- LIPSON, H. & COCHRAN, W. (1968). *The Determination of Crystal Structures*, p. 301. London: Bell.
- ROMERS, C., ALTONA, C., JACOBS, H. J. C. & DE GRAAFF, R. A. G. (1974). *Chem. Soc. Spec. Period. Rep.* **4**, Part II, ch. 3.
- VANHEE, P. (1981). To be published.
- VANHEE, P., DE PESSENIER, F., ANTEUNIS, M. & TAVERNIER, D. (1979). *Recl Trav. Chim. Pays-Bas*, **98**, 294–297.